

Vicinal $^{113}\text{Cd}, ^1\text{H}^\beta$ -Cysteine Coupling in Cd-Substituted Metalloproteins Follows a Karplus-Type Dependence

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Heteronuclear ($^{113}\text{Cd}, ^1\text{H}$) multiple quantum coherence (HMQC) has frequently been used to obtain metal–ligand connectivities in cadmium-substituted metalloproteins.^{1,2} However, no attempt has been made to obtain information concerning the position of the metal ion. Our NMR studies on ^{113}Cd -substituted metallothionein and ^{113}Cd -substituted rubredoxin indicate a Karplus-type correlation³ between the $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constants for the cysteine C^β protons and the $\text{H}^\beta\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ dihedral angle. Similar relationships have been established for vicinal couplings between lighter nuclei, such as $^1\text{H}, ^1\text{H}, ^3$ $^{15}\text{N}, ^1\text{H}, ^4$ and $^{13}\text{C}, ^1\text{H}, ^4$. In this work, we report the first demonstration of such a dependence in a system involving a heavy nucleus, i.e., $^{113}\text{Cd}, ^1\text{H}$.

Heteronuclear $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constants were determined for cysteine C^β protons in ($^{113}\text{Cd}_7$)metallothionein (Cd₇-MT) from rat liver and ^{113}Cd -substituted *Desulfovibrio gigas* rubredoxin (Cd-Rd)^{5,6} (see Table I). In both cases, the structures of the metal binding sites have been defined at high resolution by X-ray crystallography. In the case of metallothionein, a structural model of rat liver Zn₂Cd₅-MT at 2.0-Å resolution is available,⁷ and comparison with earlier NMR studies on rat liver Cd₇-MT show that both metalloforms exhibit identical molecular architectures.⁸ This protein contains three- and four-metal clusters with a total of 12 terminal and 8 bridging cysteine ligands. In this study, $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constants were extracted for terminal cysteines which bind cadmium in the crystallographically defined metalloform (see Table I). The structure of *D. gigas* rubredoxin, which contains a mononuclear Fe(CysS)₄ center, is known to 1.4-Å resolution,⁹ and we have been able to demonstrate previously that the Cd-derivative is isostructural with the native protein.¹⁰

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(5) Rat liver Zn₂Cd₅-metallothionein-2 was isolated as described in ref 13, and the $^{113}\text{Cd}_7$ -derivative was prepared as in ref 14. Heteronuclear $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constants were determined as in ref 10, using a Bruker AMX 600-MHz NMR spectrometer. Conditions used for the NMR experiments were identical to those given in ref 1b, from which the sequence-specific proton assignments were taken. The stereospecific assignments of cysteine $\text{H}^{\beta\text{a}}$ and $\text{H}^{\beta\text{b}}$ in Table I were confirmed on the basis of the $^3J(^1\text{H}^\alpha, ^1\text{H}^\beta)$ coupling constants and comparison with the crystal structure data.

(6) ^{113}Cd -substituted *D. gigas* rubredoxin was prepared as in ref 10, in which details of the sequence-specific proton assignments and the determination of the $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constants are given.

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Table I. Cysteine $^3J(^{113}\text{Cd}, ^1\text{H}^{\beta\text{a}})$ and $^3J(^{113}\text{Cd}, ^1\text{H}^{\beta\text{b}})$ Coupling Constants for Rat Liver ($^{113}\text{Cd}_7$)Metallothionein and *D. gigas* (^{113}Cd)Rubredoxin (10% D₂O/H₂O v/v, pH 7.6, 300 K)

	$\text{C}^\alpha\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ (deg)	$^3J(^{113}\text{Cd}, ^1\text{H})$ (Hz) ^{a,b}	
		$\text{H}^{\beta\text{a}}$	$\text{H}^{\beta\text{b}}$
$^{113}\text{Cd}_7\text{-MT}$			
Cys5	-46.6	51.5 (-166.6)	<1 (73.4)
Cys21	-170.6	<1 (69.4)	5.0 (-50.6)
Cys33	-81.8	47.5 (158.2)	12.5 (38.2)
Cys36	-79.9	74.3 (160.1)	10.3 (40.1)
Cys41	159.8	9.6 (39.8)	<1 (-80.2)
Cys48	-62.9	49.0 (177.1)	15.5 (57.1)
Cys57	-149.8	<1 (90.2)	19.2 (-29.8)
Cys59	-57.3	45.0 (-177.3)	4.0 (62.7)
$^{113}\text{Cd-Rd}$			
Cys6	-172.0	0.5 (68.0)	3.0 (-52.0)
Cys9	-91.0	38.0 (149.0)	17.0 (29.0)
Cys39	-178.0	0.5 (62.0)	2.5 (-58.0)
Cys42	-94.0	37.0 (146.0)	18.0 (26.0)

^a Errors are estimated to be approximately 1.5 Hz. ^b The calculated $\text{H}^\beta\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ dihedral angles (ϕ_{a} and ϕ_{b} , as illustrated in Figure 1) are given in parentheses.

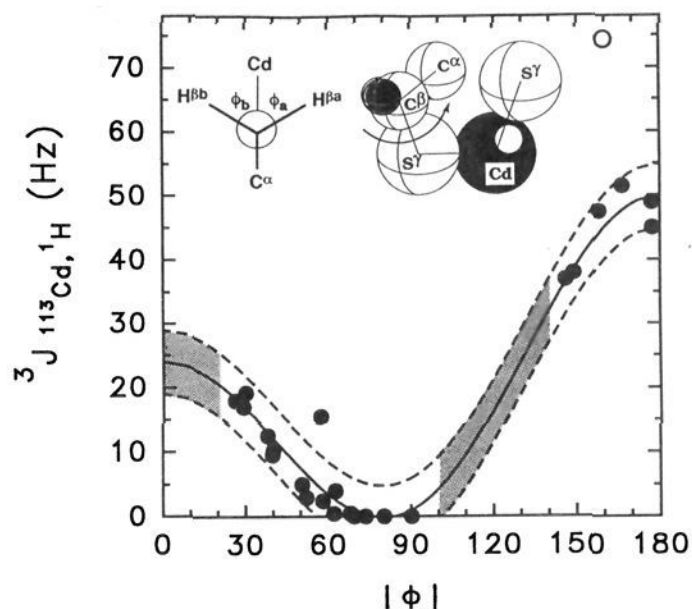


Figure 1. Correlation between the cysteine H^β $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constant and the $\text{H}^\beta\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ dihedral angle ϕ . The solid line represents a nonlinear least-squares fit of the form $c(\cos^2 \phi) - b(\cos \phi) - a$. The values calculated for the a , b , and c constants are 36, 13, and 1 Hz, respectively, with $r^2 = 98.7\%$. The dashed lines illustrate 95% confidence limits. The point shown as a hollow circle was omitted from the calculation (see text). The shaded areas represent sterically hindered conformations (see text).

Figure 1 illustrates the Karplus-type relationship obtained when the cysteine H^β $^3J(^{113}\text{Cd}, ^1\text{H})$ coupling constants are plotted against the $\text{H}^\beta\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ dihedral angles (ϕ_{a} , ϕ_{b}) (see Figure 1, inset), derived from the respective crystal structures by assuming tetrahedral geometry around C^α (see Table I). Although in general heteronuclear couplings involving heavy nuclei depend on orbital angular momentum, electron–nucleus dipole–dipole interaction, and Fermi contact terms,¹¹ it seems that, in this case, the $\text{H}^\beta\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ dihedral angle is the principal determinant in the latter term and the dominant variable influencing the system. The absence of values in the range of ϕ between approximately 100° and 140° and between 0° and 20° (shaded areas in Figure 1), corresponding to $\text{C}^\alpha\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ or $\text{H}^\beta\text{--C}^\beta\text{--S}^\gamma\text{--Cd}$ angles of less than about 20° , may be due to steric hindrance caused by the interaction of either C^α or H^β with the bulky Cd(II) ion or atoms of the adjacent cysteine ligand, resulting in these angles being strongly disfavored (see Figure 1, inset). The data point at 74-Hz coupling (see Figure 1, hollow circle), corresponding to Cys36 $\text{H}^{\beta\text{a}}$ of Cd₇-MT, clearly lies outside the expected range.

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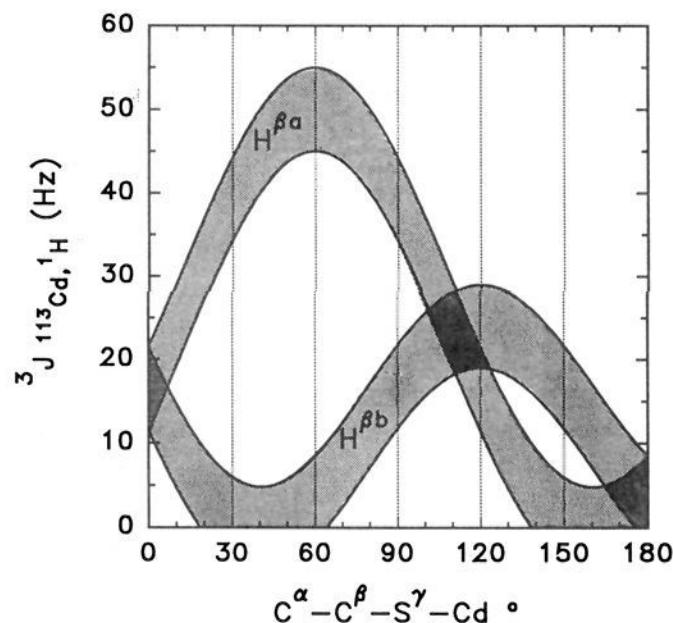


Figure 2. Expected values for the ${}^3J_{113\text{Cd}, 1\text{H}^{\beta a}}$ and ${}^3J_{113\text{Cd}, 1\text{H}^{\beta b}}$ coupling constants over the range of $\text{C}^\alpha\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ dihedral angles from 0° to 180° . As the relationship is identical for $\text{C}^\alpha\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ angles between 0° and -180° with $\text{H}^{\beta a}$ and $\text{H}^{\beta b}$ being exchanged, this region of the plot has been omitted.

As the ${}^3J_{113\text{Cd}, 1\text{H}}$ coupling constant obtained for Cys36 $\text{H}^{\beta b}$ is consistent with the relationship (see Figure 1), this unusual behavior is presumably caused by a distorted geometry around C^β .

It should be noted that each possible $\text{C}^\alpha\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ angle¹² gives rise to a unique pair of $\text{H}^{\beta a}\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ and $\text{H}^{\beta b}\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ angles and hence a unique pair of ${}^3J_{113\text{Cd}, 1\text{H}}$ coupling constants. In Figure 2, the ranges of the ${}^3J_{113\text{Cd}, 1\text{H}^{\beta a}}$ and ${}^3J_{113\text{Cd}, 1\text{H}^{\beta b}}$ coupling constants expected are defined over the range of possible $\text{C}^\alpha\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ dihedral angles. Thus, by determining the magnitudes of these coupling constants, an estimate of the $\text{C}^\alpha\text{-C}^\beta\text{-S}^\gamma\text{-Cd}$ angle can be obtained. This relationship will clearly be of great use in defining accurately the geometries of metal binding sites in ${}^{113}\text{Cd}$ -substituted metalloproteins, such as zinc-finger proteins, which contain cysteine ligands. Such information may prove useful as an additional input parameter in three-dimensional structure analysis by NMR methods, permitting a more precise determination of the position of the metal ion.

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